metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Guang-Bo Che* and Chun-Bo Liu

Department of Chemistry, Jilin Normal University, Siping 136000, People's Republic of China

Correspondence e-mail: guangbochejl@yahoo.com

Key indicators

Single-crystal X-ray study T = 292 KMean $\sigma(\text{C}-\text{C}) = 0.006 \text{ Å}$ R factor = 0.064 wR factor = 0.138 Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A new dinuclear Mn^{II} compound with intermolecular $\pi - \pi$ interactions

In the title compound, bis(μ -pyridine-2,5-dicarboxylato)bis-[aqua(pyrazino[2,3-*f*][1,10]phenanthroline)manganese(II)] dihydrate, [Mn₂(C₇H₃NO₄)₂(C₁₄H₈N₄)₂(H₂O)₂]·2H₂O, each Mn^{II} atom is six-coordinated by three N atoms from one bidentate pyrazino[2,3-*f*][1,10]phenanthroline and one monodentate pyridine-2,5-dicarboxylate ligand, and three O atoms from two monodentate pyridine-2,5-dicarboxylate ligands and one water molecule. The two Mn^{II} atoms are bridged by two pyridine-2,5-dicarboxylate ligands, forming a centrosymmetric dinuclear molecule. Neighboring molecules interact through π - π contacts, leading to a one-dimensional supramolecular structure, and O-H···O hydrogen bonds complete the structure.

Comment

1,10-Phenanthroline (phen) and its derivatives are important ligands with numerous uses in the construction of metalorganic complexes (Che, Liu *et al.*, 2006). Supramolecular architectures based on the pyrazino[2,3-*f*][1,10]phenanthroline (Pyphen) molecule have received considerably less attention (Che, Xu & Liu, 2006). We selected pyridine-2,5dicarboxylic acid (2,5-H₂pdc) as a linker ligand and Pyphen as a secondary ligand, generating a new dinuclear coordination compound, $[Mn_2(2,5-pdc)_2(Pyphen)_2(H_2O)_2]\cdot 2H_2O$, (I), which is reported here.



In (I), each Mn^{II} atom is six-coordinated by three N atoms from one bidentate Pyphen molecule and one monodentate 2,5-pdc ligand, and three O atoms from two monodentate 2,5pdc ligands and one water molecule. The Mn—N and Mn—O distances are normal (Table 1): a disrorted octahedral *mer*-MnO₃N₃ arrangement is formed. The complete molecule contains two Mn^{II} atoms bridged by two 2,5-pdc ligands, forming a dinuclear compound (Fig. 1), the complete assemblage being generated by inversion symmetry. The carboxylate C—O bond lengths suggest that the bonding of the C15/O1/O2 group is localized, but that of the C21/O3/O4 group is delo-

© 2006 International Union of Crystallography All rights reserved Received 27 June 2006 Accepted 27 June 2006





View of the structure of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level (arbitrary spheres for the H atoms). [Symmetry code: (i) 1 - x, 1 - y, -z.] The symmetry-related uncoordinated water molecule has been omitted.



Figure 2

View of the one-dimensional supramolecular structure of (I), generated by π - π interactions. H atoms and water molecules have been omitted.

calized. Despite this, the Mn1-O1 and $Mn1-O4^{i}$ (see Table 1 for symmetry codes) bond lengths are very similar.

In the crystal structure, neighboring molecules are connected through π - π interactions between Pyphen ligands with a stacking distance of 3.47 Å, resulting in a one-dimensional supramolecular structure (Fig. 2). Various O-H···O hydrogen bonds involving the water molecules and carboxylate O-atom acceptors (Table 2) complete the structure of (I).

Experimental

The Pyphen ligand was synthesized according to the literature method of Dickeson & Summers (1970). A methanol solution (10 ml) of Pyphen (0.5 mmol) was added slowly to an aqueous solution (8 ml) of $MnCl_2 \cdot H_2O$ (0.5 mmol) and 2,5-H₂pdc (0.5 mmol) with stirring. The resulting solution was filtered and the filtrate was allowed to stand in air at room temperature for several days, yielding pale-yellow crystals of (I) (39% yield based on Mn).

Crystal data

$\gamma = 109.47 \ (3)^{\circ}$
V = 970.6 (4) Å ³
Z = 1
$D_x = 1.671 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
$\mu = 0.73 \text{ mm}^{-1}$
T = 292 (2) K
Block, pale yellow
$0.33 \times 0.27 \times 0.21 \text{ mm}$

Data collection

```
Rigaku R-AXIS RAPID
diffractometer
```

Absorption correction: multi-scan (ABSCOR; Higashi, 1995)

 $T_{\min} = 0.784, \ T_{\max} = 0.852$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.064$ $wR(F^2) = 0.138$ S = 1.054413 reflections 310 parameters H atoms treated by a mixture of independent and constrained refinement 9612 measured reflections 4413 independent reflections 2620 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.081$ $\theta_{\text{max}} = 27.5^{\circ}$

$$\begin{split} &w = 1/[\sigma^2(F_{\rm o}^{\,2}) + (0.05P)^2 \\ &+ 0.1792P] \\ &where \ P = (F_{\rm o}^{\,2} + 2F_{\rm c}^{\,2})/3 \\ (\Delta/\sigma)_{\rm max} < 0.001 \\ \Delta\rho_{\rm max} = 0.39 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.54 \ {\rm e} \ {\rm \AA}^{-3} \end{split}$$

Table 1			
Selected	geometric parameters	(Å,	°).

Mn1 - O1W	2.110 (3)	Mn1-N1	2.324 (3)
Mn1-O1	2.156 (3)	C15-O2	1.236 (4)
Mn1-O4 ⁱ	2.143 (3)	C15-O1	1.280 (4)
Mn1-N2	2.241 (3)	C21-O4	1.248 (5)
Mn1-N5 ⁱ	2.278 (3)	C21-O3	1.252 (4)
$O1W-Mn1-O4^{i}$	166.72 (12)	$O1-Mn1-N5^{i}$	92.22 (11)
O1W-Mn1-O1	90.57 (11)	N2-Mn1-N5 ⁱ	161.67 (10)
O4 ⁱ -Mn1-O1	96.89 (11)	O1W-Mn1-N1	85.04 (12)
O1W-Mn1-N2	102.65 (12)	O4 ⁱ -Mn1-N1	90.46 (11)
O4 ⁱ -Mn1-N2	87.86 (11)	O1-Mn1-N1	164.46 (11)
O1-Mn1-N2	93.72 (11)	N2-Mn1-N1	72.82 (12)
O1W-Mn1-N5 ⁱ	94.60 (12)	$N5^i - Mn1 - N1$	102.97 (12)
O4 ⁱ -Mn1-N5 ⁱ	74.22 (11)		

Symmetry code: (i) -x + 1, -y + 1, -z.

l able 2				
Hydrogen-bond	geometry	(Å,	°)	

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
	$\begin{array}{c} O1W - HW11 \cdots O2W \\ O1W - HW12 \cdots O3^{ii} \\ O2W - HW21 \cdots O1^{iii} \\ O2W - HW22 \cdots O2^{iv} \end{array}$	0.83 (4) 0.85 (2) 0.85 (2) 0.85 (2)	1.89 (2) 1.82 (2) 2.04 (2) 1.93 (2)	2.704 (4) 2.665 (4) 2.874 (4) 2.767 (4)	166 (4) 176 (4) 165 (4) 170 (4)

Symmetry codes: (ii) -x + 2, -y + 1, -z; (iii) -x + 1, -y, -z; (iv) x + 1, y, z.

All H atoms on C atoms were positioned geometrically (C–H = 0.93 Å) and refined as riding, with U_{iso} (H)= $1.2U_{eq}$ (C). The H atoms of water molecules were located in difference Fourier maps and their positions and U_{iso} values were refined freely.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *PROCESS-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Sheldrick, 1990); software used to prepare material for publication: *SHELXL97*.

The authors thank Jilin Normal University for supporting this work.

References

Che, G.-B., Liu, C.-B., Cui, Y.-C. & Li, C.-B. (2006). Acta Cryst. E62, m1362m1364.

Che, G.-B., Xu, Z.-L. & Liu, C.-B. (2006). Acta Cryst. E62, m1370–m1372. Dickeson, J. E. & Summers, L. A. (1970). Aust. J. Chem. 23, 1023–1027.

Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.

Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.

- Sheldrick, G. M. (1990). SHELXTL-Plus. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.